<u>Unit 2 Vocabulary Definitions</u>

Directions: Copy the following definition *exactly as written* into your vocabulary chart. Then use definition to finish the other boxes.

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1	Acid	Substances that dissolve in water to yield hydrogen ions and produce a
		solution with a pH less than 7.
2	Activation Energy	The minimum energy required for a chemical reaction to occur.
3	Adhesion	The joining of molecules of two different substances due to attractive
		forces.
4	Atom	The smallest unit of an element having all the characteristics of that
		element.
5	Base	Substances that dissolve in water to yield hydroxyl ions and produce a
		solution with a pH greater than 7.
6	Buffer	A substance that minimizes the change of the acidity of a solution when an
		acid or base is added.
7	Carbohydrate	Any of the group of organic compounds consisting of carbon, hydrogen,
	carbonyarate	and oxygen, usually in the ratio of 1:2:1.
	Carbon Compound	
	carbon compound	Compounds that contain carbon atoms bonded to other carbon atoms.
	Cohesion	
	Collesion	The joining of molecules of the same substance due to attractive forces.
10	Company	
	Compound	A substance made up of two or more elements that are chemically bonded.
		The lightweight subatomic particle that carries a negative charge and is
11	Electron	found in the electron cloud.
12		Any of the 116 known substances that cannot be separated into smaller
	Element	substances.
		A protein that speeds up chemical reactions in the body by acting as a
13	Enzyme	
		catalyst and lowering the temperature needed for the reaction to take
		place.
14	Lipid	The group of fatty or fatlike substances that are insoluble in water.
	Macromolecule	A large molecule consisting of smaller units linked together.
16	Neutron	A subatomic particle with no electrical charge found in the nucleus of an
		atom.
17	Nucleic Acid	The class of substances that include DNA and RNA and are located within
		the cells of all living things.
18	рН	
		The number that indicates the relative acidity or alkalinity of a substance.
	Protein	Large organic compounds that are composed of amino acids held in
		peptide bonds to form polypeptides.
	Proton	The subatomic particle with a positive charge found in the nucleus of an
	1 1 0 1 0 1	atom.